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A General Pressure Generation Model for Granular Propellant Fires

Frederick Paquet, Hoi Dick Ng and Mario Paquet



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Solid propellants transform chemical enegy into mechanical energy through the generation of high pressures caused by the completion of a combustion reaction in a limited volume.



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The high pressures are caused by:

- A solid to gas transformation in a limited volume
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Since applications vary widely in their requirements, propellants can exhibit a large array of combustion behaviors.



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Previous work

An attempt was made by Graham to model explosion venting and define a critical vent area ratio by equating pressure rise and decrease terms:

 $\frac{A_v}{S_B} = \frac{RT_B\rho\alpha}{MC_DA^*(A-BT_o)}$



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A number of empirical relations have been published for gas and dust equations (see NFPA 68 for examples).

• Gas explosions:
$$A_v = CA_s P_{red}^{-1/2}$$

• Dust Explosions:
$$A_v = 10^{-4} K_{st} V^{0.75} \left(\frac{P_{max}}{P_{red}} \right)^{1/2}$$

 \rightarrow Deflagration index, K_{st} , and reduced pressure, P_{red} .

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Theoretical background

Equation of state:

Nobel-Abel:

P(V-b) = NRT

when $X = cb \gg 0.01$

Most cases can use the ideal gas law:

$$PV = NRT$$

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Equation of state:

Spatial pressure variation:

Nobel-Abel:

$$\chi = rac{m_{air}c_{air}}{\dot{m}_{gen}d}$$

P(V-b) = NRT

when

Lumped parameters $\rightarrow \chi \gg 1$

 $X = cb \gg 0.01$ Most cases can use

the ideal gas law:

Case	χ
	(dimensionless)
0.7 L closed vessel with a fast burning propellant	0.9
0.7 L closed vessel with a slow burning propellant	5.6
60 L tank with a fast burning propellant	420
1800 L enclosure with a fast burning propellant	561
106 L enclosure with a fast burning propellant	675

PV = NRT

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Theoretical background Compressibility: Equation of state: Spatial pressure variation: Studied pressures below 10 kPa Nobel-Abel: $\chi = \frac{m_{air}c_{air}}{\dot{m}_{gen}d}$ P(V-b) = NRT $\rightarrow Ma < 0.30$ Lumped parameters $\rightarrow \chi \gg 1$ Bernoulli's law: when $X = cb \gg 0.01$ $\mathbf{v} = \left(\frac{2P}{\rho_{gas}}\right)^{1/2}$ Case (dimensionless) Most cases can use 0.7 L closed vessel with a fast burning propellant 0.9 0.7 L closed vessel with a slow burning propellant 5.6 the ideal gas law: 60 L tank with a fast burning propellant 420 Other cases: 1800 L enclosure with a fast burning propellant 561 106 L enclosure with a fast burning propellant 675 isentropic eq.

PV = NRT

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Theoretical model

Mass balance: $\dot{m}_e = \dot{m}_{gen} - \dot{m}_{vent}$

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Vented gases can be modelled by Bernoulli's law and the ideal gas law:

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The end result is the following model:

$$\dot{m}_e + C_D A igg(rac{2
ho_{gas}RT}{M_w V} igg)^{1/2} m_e^{1/2} - \dot{m}_{gen} = 0 \qquad \textit{when} 0 \leq t \leq t_{\textit{burn}}$$

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Testing





1800 L setup

Typical measurement

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Testing

Propellant	Geometry	Heat of explosion	Composition	
		(J/kg)		
SB1	Unitubular	3871	NC: 98% / Inert: 2%	
SB2	Unitubular	3135	NC: 90% / Inert: 10%	
DB1	Cord	5392	NC: 60% / NG: 39% / Inert: 1%	
DB2	Unitubular	4490	NC: 73% / NG: 25% / Inert: 2%	





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Flame propagation – Induction time



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Data analysis

Best statistical model: $P_{max} = \frac{194.6m^{0.97}}{A^{1.34}}$ (with $r^2 = 0.91$)



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Departure from theoretical form suggests: $C_D \approx A^{1/2}$

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Starting with the previously derived mass balance:

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An energy balance is also considered:

$$\Delta T = \frac{\dot{m}_{gen} E \Delta t - C_v \dot{m}_{vent} T \Delta t}{C_v (m_{air} + \rho_{gas} V)}$$

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Here, the mass generation rate can be approximated by:



Volume	Min. density	Max. Temperature
(m ³)	(kg/m^3)	(K)
0.06	0.74	452
1.8	0.48	823
100	0.18	1798

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The gas density is estimated as

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$$ho_{gas} = 0.47 \, V^{-0.19}$$

Subst. in theoretical model

e.g.
$$P_{max} = \frac{1.06\dot{m}^2 V^{0.19}}{A^{2.66}}$$

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Comparing the scale dependent model and measured results:



 \rightarrow Loss in quality due to the uncertainty in estimating \dot{m}_{gen} .

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Future work

- Application to safety aspects
 - Quantities distances relations, uniforms, building process design, ...
- Better estimate of the mass combustion rate
- Application to "non dry" (or green) product
- Application to stick geometries (e.g. rocket motors)
- Application to cases at higher densities (higher pressures)



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