



# U.S. ARMY COMBAT CAPABILITIES DEVELOPMENT COMMAND – ARMAMENTS CENTER

## Continuous Reactors for Energetic Materials Synthesis

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Research Scientist III  
CS Squared, LLC

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# AGENDA



- **Answering the “When” of Flow**
  - Advantages of Flow Chemistry: The Three C’s
    - Chemistry
    - Control
    - Cost
  - Screening Reactions for Continuous
    - Calabrese and Pissavini Screening Framework
    - Roberge Reaction Classification Scheme
    - Reaction/Reactor Selection Matrix
  
- **Answering the “What” of Flow**
  - Transformers and Generators
  - Survey of Commercial Off-The-Shelf Apparatus
    - Continuous Reactors
    - Continuous Crystallizers and Filters
    - Ancillary Equipment
      - Spectroscopic Tools
      - Pumps
      - Separators
  
- **Summary and Additional Resources**



# Answering the “When” of Flow



# ADVANTAGES OF FLOW: THE THREE C's



- **Chemistry**

- Safety: Smaller scale; No headspace; No accumulation of reactive or toxic intermediates
- Expansion of the Reaction Space: “Forbidden” reactions are feasible in flow; Reactions in Novel Process Windows (high temperature, high pressure, increased concentration/solvent-free, explosive or thermal runaway regimes)

- **Control**

- Seamless Scale Up: Scale up (numbering up) is faster and more reliable
- Versatility and Flexibility: Customizable and adjustable equipment; Ease of switching products; Flexibility to turn throughput up or down
- Increased space–time yield (increased throughput in smaller footprint)
- Steady-state, continuous process feedback
- Quality by Design (QbD)/Quality by Control (QbC)

- **Cost**

- Less energy, water, and raw material costs (solvents, cleaning fluid)
- Lower labor cost
- Less capital depreciation
- Increased product yield and quality



# SCREENING REACTIONS FOR CONTINUOUS



|   |  |                               |
|---|--|-------------------------------|
| Pressures exceed reactor capability?            | Likely no benefit and/or not possible            | Increasing applicability<br>↓ |
| Temperatures exceed reactor capability?         |  |                               |
| Chemistry not compatible with reactor?          |  |                               |
| Equilibrium reactions?                          |  |                               |
| Solid precipitates?                             | Possible benefits, but with technical challenges |                               |
| Very slow kinetics?                             |  |                               |
| Solid reactants or catalysts?                   |  |                               |
| Gaseous reactants?                              |  |                               |
| Homogeneous catalysts?                          |  |                               |
| Gas evolution?                                  | Likely to benefit                                |                               |
| Reaction benefits from pressure > 120 Pa        |  |                               |
| Unstable intermediates?                         |  |                               |
| Fast kinetics?                                  |  |                               |
| Highly toxic reactants or byproducts?           |  |                               |
| Reactions requires or benefits from T < -10 °C? |  |                               |
| Rapid mixing required?                          |  |                               |
| Highly exothermic?                              |  |                               |
| Over-reaction possible?                         |  |                               |
| Requires precise stoichiometric control?        |  |                               |



# ROBERGE REACTION CLASSIFICATION SCHEME



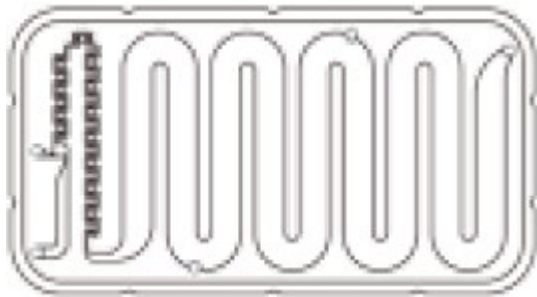
- **Type A reactions**
  - Very fast, typically  $< 1$  s
  - Controlled by diffusion and mixing
  - Improved mixing/heat exchange may increase yield
- **Type B reactions**
  - Rapid reactions, typically 10 s to 30 min
  - Kinetically controlled – rate may be accelerated by increased temperature, pressure, or concentration
- **Type C reactions**
  - Slow reactions,  $> 30$  min to hours
  - Involve potential hazards such as autocatalysis or thermal accumulation
  - Kinetics make reactions suitable for batch, but continuous offers improved safety or product quality
- **Type D reactions**
  - Reactions that are not A, B, or C
  - Should be intensified to at least Type C to be done in flow



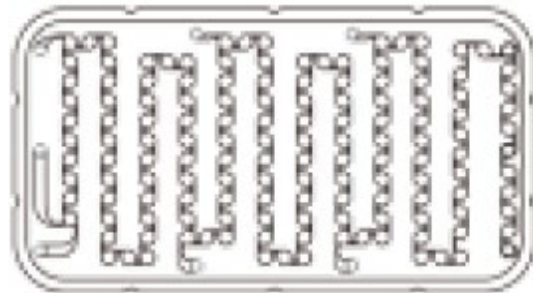
# REACTION/REACTOR MATRIX



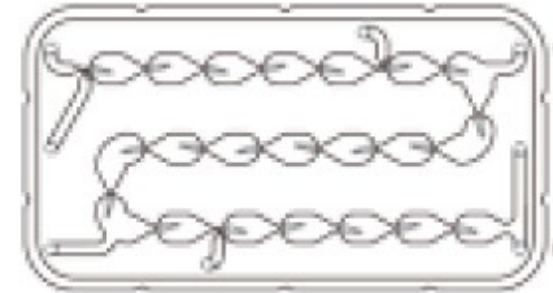
| Rates/Phases | Homogeneous          | Liquid-Liquid                | Gas-Liquid                    | Solid-Liquid        |
|--------------|----------------------|------------------------------|-------------------------------|---------------------|
| Type A       | Plate SZ/TG          | Plate LL                     | Plate LL                      | CSTR/<br>Packed Bed |
| Type B       | Plate SZ/TG<br>Coil  | Plate LL<br>Coil pulsated    | Plate LL<br>Coil pressure     | CSTR/<br>Packed Bed |
| Type C       | Static mixer<br>Coil | State mixer<br>Coil pulsated | Static mixer<br>Coil pressure | Coil pulsated       |



**SZ**  
**(serpentine)**



**TG**  
**(tangential)**



**LL**  
**(liquid-liquid)**



# Answering the “What” of Flow





# TRANSFORMERS AND GENERATORS



## Transformers

### Oxidation Modules

- O<sub>2</sub> via two-feed approach
- Tube in tube reactor
- Homogeneous oxidant & T-mixer

### Reduction Modules

- H<sub>2</sub> via tube in tube combined with catalyst
- *In situ* H<sub>2</sub> generation—pressurized gas-liquid mixer & packed bed reactor
- Homogenous reductions (DIBALH)

### Nitrations

- Mixed acid nitrations
- Nitration using acetyl nitrate (HNO<sub>3</sub>-Ac<sub>2</sub>O-AcOH)
- Alternative Nitration Condition (N<sub>2</sub>O<sub>5</sub> + TFE; NH<sub>4</sub>NO<sub>3</sub>-H<sub>2</sub>SO<sub>4</sub>)

### Olefinations

- Polymer-assisted HWE olefination using packed bed reactor
- Wittig reaction using SS coil at 210 °C

## Generators

### *In situ* Hazardous Materials

- Stable Reagents
  - Cl<sub>2</sub>, Br<sub>2</sub>, BrCN, HCN, NOCl, N<sub>2</sub>H<sub>4</sub>, SO<sub>2</sub>, COCl<sub>2</sub>, ClCH<sub>2</sub>OCH<sub>3</sub>, R-NC
- Limited Stability Reagents
  - H<sub>2</sub>O<sub>2</sub>, RCO<sub>3</sub>H, N<sub>2</sub>CHCO<sub>2</sub>Et, R-N<sub>3</sub>, LDA
- Unstable Reagents
  - O<sub>3</sub>, CH<sub>2</sub>N<sub>2</sub>, CF<sub>3</sub>CHN<sub>2</sub>, HN<sub>3</sub>, BrN<sub>3</sub>, ClN<sub>3</sub>, <sup>1</sup>O<sub>2</sub>

### Reactive Intermediates

- Iminium cations
- Benzyne
- R-MgX
- Aryllithium
- Carbon-centered radicals
- Alkyl diazo compounds
- Benzyl diazo compounds



# Continuous Reactors for Energetic Materials Synthesis

Commercial-Off-The-Shelf Apparatus



# LABORATORY REACTORS



## Uniqsis FlowSyn

[www.uniqsis.com](http://www.uniqsis.com)

- All PTFE or PTFE–Hastelloy construction available
  - Temperature:  $-70$  to  $+260$  °C; Pressure: 100 bar
- Data Logger and FlowControl™ Software
  - Plan experiments; Monitor and log temperature, pressure, and flow rate
  - Wireless or remote control over LAN
- Up to 3 temperature zones and 3 reagent inputs available
- Available LED Photoreactor

## Vapourtec RS-400



- PFA or Hastelloy construction available
  - Temperature:  $-70$  to  $+250$  °C; Pressure: 50 bar
- Flow Commander™ Software
  - Plan experiments; Monitor and log temperature, pressure, and flow rate
  - API Package can be driven by .NET Software Framework
- Up to 4 reactors and 4 reagent inputs available
- Available Photoreactor (Hg lamp or LED) and Electrochemical reactor

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## Ehrfeld Modular MicroReaction System (MMRS)



- Hastelloy construction available
  - Temperature:  $-20$  to  $+200$  °C ( $-100$  to  $+600$  °C); Pressure: 100 bar
- Integrates with LabManager® automation system from HiTec Zang
  - Consists of a control unit and LabVision® visualization and automation software
- More than 60 different microreaction modules
- Available Photoreactor (UV lamp or LED)

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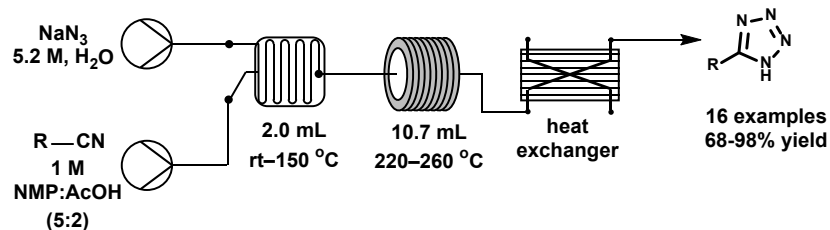


# LABORATORY REACTORS



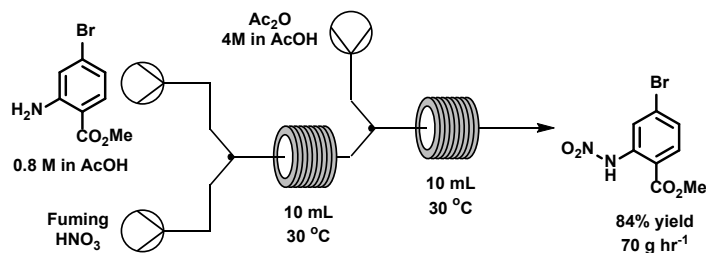
## Uniqsis FlowSyn

www.uniqsis.com



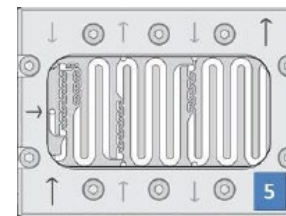
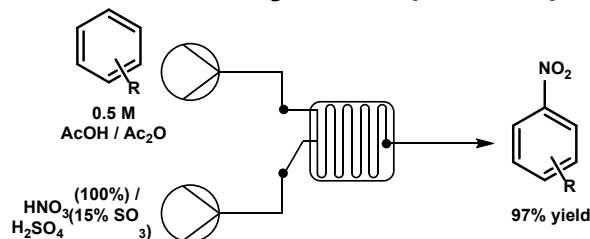
Gutman, B. et al. *Angew. Chem. Int. Ed.*,  
**2010**, 49, 7101-7105.

## Vapourtec RS-400



Brocklehurst, C. E.; Lehmann, H.; La Vecchia, L. *Org. Process Res. Dev.*, **2011**, 15, 1447-1453.

## Ehrfeld Modular MicroReaction System (MMRS)



LL-Mixer 0.24 mL internal volume

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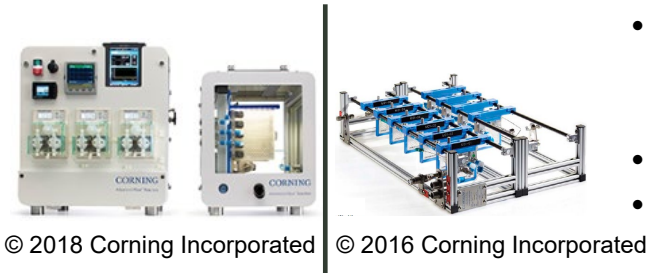
Köckinger, M. et al. *Org. Process Res. Dev.*, **2020**, 24, 2217-2227.  
 Sagmeister, P. et al., *React. Chem. Eng.*, **2020**, 5, 677-684.



# SCALABLE REACTORS



## Corning Advanced-Flow™ Reactors

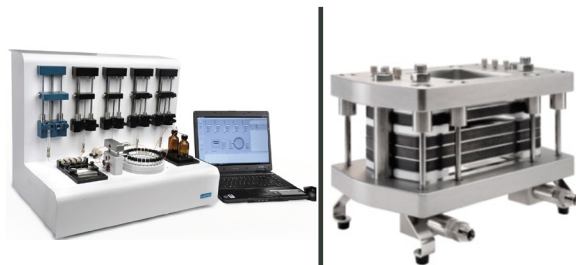


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- Borosilicate glass or SiC (G1 and larger) construction available
  - Temperature:  $-60$  to  $+200$  °C; Pressure: 18 barg
  - Metal-free system for high chemical durability
- Integrated thermostat; plug and play system with data monitoring
- Lab Photo Reactor Option
  - Tunable LED source with 6 different wavelengths; wireless control

## Chemtrix Labtrix® S1 and Protrix®



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- Inert wetted materials:
  - Labtrix® S1: PTFE, ETFE, FFKM, Glass
  - Protrix®: PTFE, FFKM, SiC
- Temperature:  $-20$  to  $+195$  °C; Pressure: 20 bar
- Volume: 1 to 19.5  $\mu$ L (Labtrix® S1); 1 to 13.5 mL (Protrix®)
- Dedicated software for automated data logging and sample collection
- Protrix® processes can be directly scaled to production (Plantrix®)

## AM Technology Coflore® ACR



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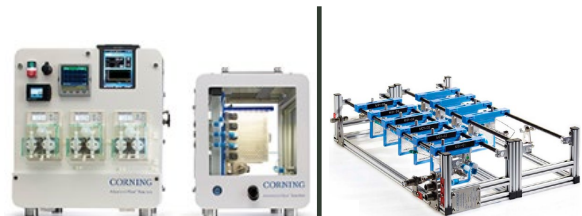
- Hastelloy reactor block; Hastelloy or ceramic agitators
  - Temperature:  $-40$  to  $+140$  °C; Pressure: 10 bar
- ACR cell block has 10 reaction cells connected via interstage channels
  - Sample points, addition points, & temp measurement can be added to any reaction cell
- Agitating platform provides horizontal agitation to reactor block for mixing
- Temperature data & agitation rates can be accessed by USB or LAN



# SCALABLE REACTORS

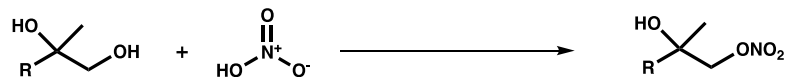


## Corning Advanced-Flow™ Reactors

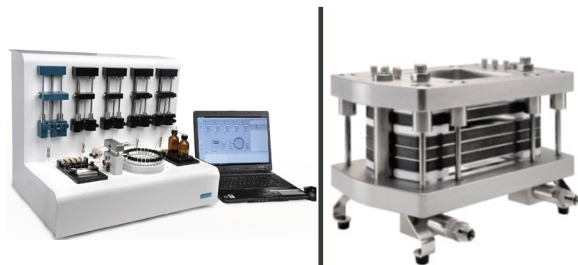


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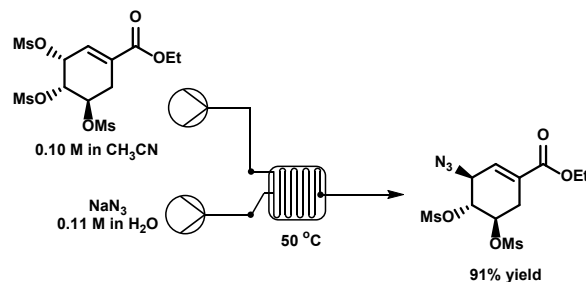
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Braune, S. *et al. Chemistry Today*, **2008**, 26(5), 1-4.

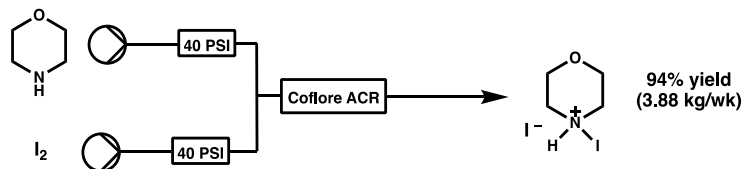
## Chemtrix Labtrix® S1 and Protrix®



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Sagandira, C. R.; Watts, P. *Beilstein J. Org. Chem.*, **2019**, 15, 2577-2589.

## AM Technology Coflore® ACR

Browne, D. L.; Deadman, B. J.; Ashe, R.; Baxendale, I. R.; Ley, S. V. *Org. Process Res. Dev.*, **2011**, 15, 693-697.

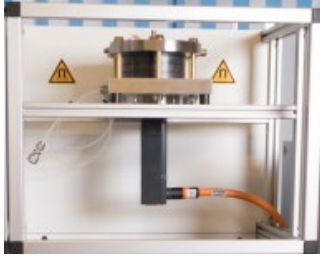
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# SPINNING DISC REACTORS



## Flowid SpinPro R10



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- SiC Spinning Disc Reactor
  - Three-stage reactor; 19 mL total volume
  - Temperature:  $-20$  to  $+160$  °C; Pressure: 10 bar
- Discs can be mechanically or chemically modified
- Suitable for precipitations and for controlled emulsification
- Pilot (R300) and production (R1000) scale units available

## KinetiChem Synthetron™



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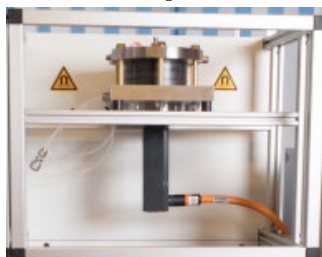
- Hastelloy C-22 surfaces; inert fluoropolymer seals
  - 10  $\mu$ L to 1.3 mL lab scale spinning disc reactor
  - Temperature:  $-40$  to  $+150$  °C; Pressure: 10 bar
- 1,000 W motor; speed up to 14,000 RPM
- TouchScreen Allen Bradley Micro800 series controller/data logger
  - 4 x K-type thermocouple collection ports
  - 2 x 4-20 mA pressure transmitters



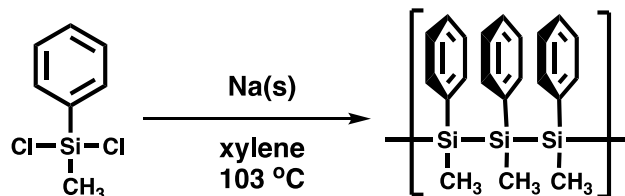
# SPINNING DISC REACTORS



## Flowid SpinPro R10



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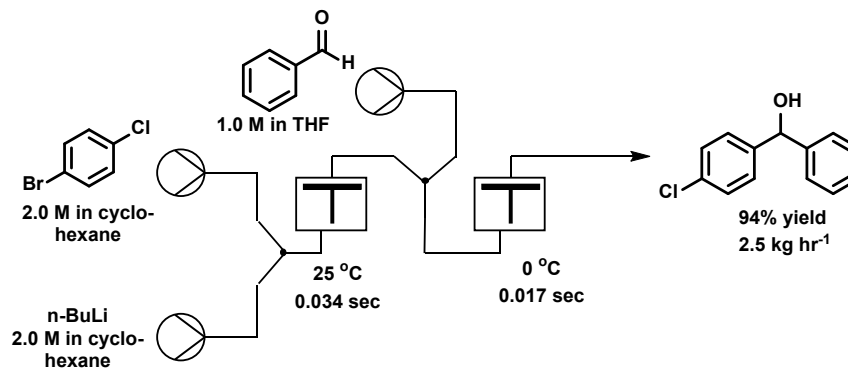


Hees, M.; Georgi, U.; Bachus, H.; Muller, K-S. US Patent Office US 2018/0346655, December 6, 2018

## KinetiChem Synthetron™



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Slocum, D. W. *et al.*, *Tetrahedron Lett.*, **2010**, 51, 4793-4796.





# NITECH® SOLUTIONS

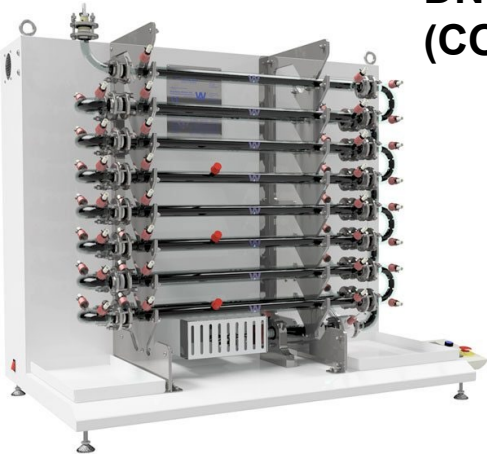


## DN25 Oscillatory Baffled Reactor (OBR)

- Borosilicate glass vessel; 25 mm diameter; 110 mL internal volume
- Temperature:  $-20$  to  $+120$  °C; Pressure: ambient
- Oscillator Frequency: 0.1 to 3 Hz in 0.1 Hz increments
- Oscillator Stroke: 5 to 40 mm in 1 mm increments
- May be configured for batch, semi continuous, or continuous operation
- Optimized OBR parameters may be used for COBC

## DN6/DN15 Continuous Oscillatory Baffled Crystallizer/Reactor (COBC/R)

- Glass (DN6/DN15) or Hastelloy (DN15); 200 mL (DN6)/1.25 L to 4.5 L (DN15) internal volume
- Temperature: 0 to  $+100$  °C (DN6);  $-20$  to  $+150$  °C (DN15)
- Pressure: 0 to 3 bar (DN6); 0 to 10 bar (DN15)
- Oscillator Frequency: 0.1 to 6 Hz (DN6); 0.1 to 3 Hz (DN15)
- Oscillator Stroke: 2 to 25 mm (DN6); 11 to 68 mm (DN15)
- ATEX compliant DN15 models available

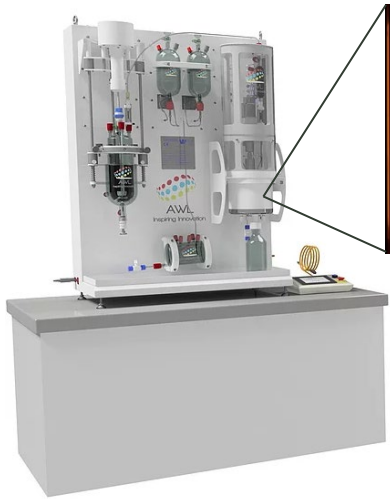


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Peña, R.; Olivia, J. A.; Burcham, C. L.; Jarmer, D. J.; Nagy, Z. K. *Cryst. Growth Des.*, **2017**, *17*, 4776-4784.  
Kacker, R.; Maaß, S.; Emmerich, J.; Kramer, H. *AlChE J.*, **2018**, *64*, 2450-2461.



# ALCONBURY WESTON CCF50 AND CCF20-LITE CONTINUOUS CAROUSEL FILTER



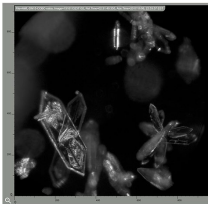
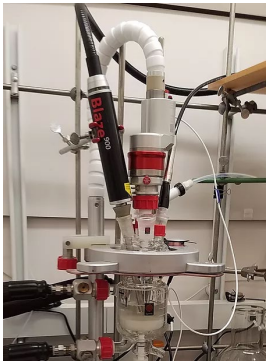
- **CCF50 For Hazardous Locations:**
  - Meets DOE Standard 1212 for Energetic Materials
  - No/Minimal particle shear
  - Up to three wash solvent reservoirs
  - 1 moving part
  - 50 mm ports – 40 g of material per port
  - Entry-to-exit time typically 10 to 15 minutes
  - Cake wetted to set parameters before discharge
  - Optional dryer
    - Dry gas fed to mass spec to monitor solvent content



- **CCF20-Lite Features:**
  - New product provides CCF20 functionality w/o ancillaries
  - Auto transfer function from external reactor/crystallizer
  - Automated filtration and wash cycle
  - Optional N<sub>2</sub> blanket system
  - Optional cooling vessel for wash solvent



# BLAZE METRICS™ AND TORNADO SPECTRAL SYSTEMS

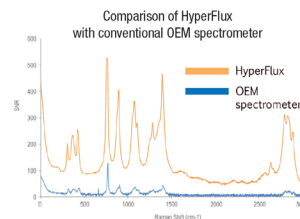


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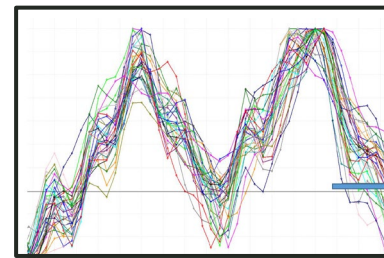
## One Probe with Multiple Integrated Technologies – Simultaneous Acquisition:

- **Microscopy:** high contrast, high resolution, highest dynamic range for understanding single dispersed phase and multiple phase, multiple component particle systems
- **High Dynamic Range Turbidity (HDR):** dynamic range measuring from low to extremely high dispersed phase concentrations; can track change at nano or micron scale, measure optical transitions in liquid and/or solid phase, operate in translucent to black solutions
- **Advanced Chord Length (A-CLD):** removes flow speed artifacts and reduces multiple other artifacts of scanning tools; track changes in particle size, count, and shape
- **Particle Focused Raman (PFR):** can dramatically increase Raman signal captured from dispersed phase particles; find and track polymorphs, solvates, hydrates, impurities; differentiate multiple component systems.
- **Immersed Probe Tip Material:** Hastelloy 22, 276, SS, or custom
- **Window Materials:** Sapphire, Kalrez, Nickel, and Gold plate
- **Temp:** -10 to 100 °C (-10 °C requires N<sub>2</sub> purge)
- **Pressure:** 6 bar (standard); 22 bar (optional)

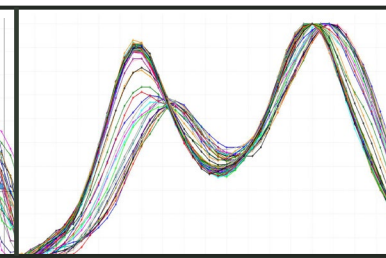
- **HyperFlux™ Pro Plus Raman**
  - 200–3300 cm<sup>-1</sup> optical range
  - Laser Power: 20 mW to 495 mW (ATEX compliant)
  - Excitation wavelength: 785 nm



Ball Probe w/Tornado  
(pre-processed)



Blaze PFR Probe  
w/Tornado (raw data)



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# MAGRITEK BENCHTOP NMR



## • Available Pulse Sequences

### • Proton

- 1D
- Paramagnetic
- 2D COSY
- 2D TOCSY
- 2D JRES
- $T_1$ ,  $T_2$
- Reaction Monitoring

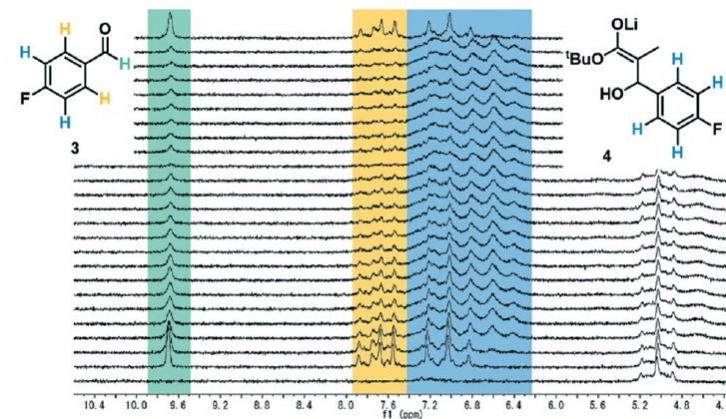
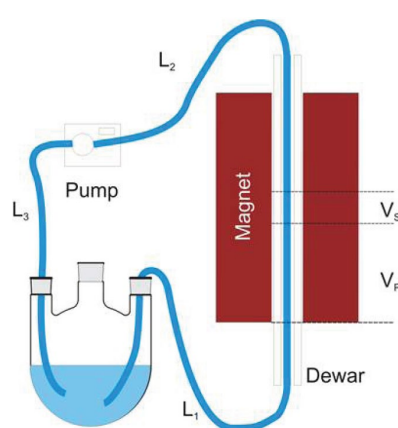
### • Carbon

- 1D
- DEPT
- HETCOR
- HMBC
- HMQC
- HSQC
- HSQC-ME

## • Specifications

- 43, 60, or 80 MHz ( $^1\text{H}$ )
- 23" x 17" x 16"
- 120 lbs
- 110–240 AC, 60 Hz
- Operating Temp: 20 to 25 °C
- 2 G line completely inside spectrometer

- Optional third nuclei:  $^7\text{Li}$ ,  $^{11}\text{B}$ ,  $^{15}\text{N}$ ,  $^{23}\text{Na}$ ,  $^{29}\text{Si}$ ,  $^{19}\text{F}$ , and  $^{31}\text{P}$





## PUMPS



### Vapourtec SF-10



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- Self-priming
- Flow rate: 0.02 to 10 mL min<sup>-1</sup>
- Max pressure: 10 bar
- Pumps solutions, suspensions, light slurries, and gasses
- Versatile options for external control
- Operating Modes:
  - Constant flow rate or ramped flow rate
  - Volume dosing
  - Gas delivery
  - Pressure controller

### Fuji Techno SMP

[www.fuji-techno.co.jp/english](http://www.fuji-techno.co.jp/english)

- Super Metering Pump is pulse-free; metal-free version available
- Able to feed liquid at  $\pm 0.1\%$  of specified flow range
- Max flow rate: 15.3 to 108.6 mL min<sup>-1</sup>
- Max discharge pressure: 20 bar
- Applications
  - Additive feed into extruder
  - High precision dosing
  - Line mixing
  - Emulsification

### Teledyne ISCO



© Teledyne ISCO. All rights reserved.

- Seven D-series pumps
  - Flow rates:  $< 1 \mu\text{L min}^{-1}$  to 400 mL min<sup>-1</sup>
  - Max pressure:  $> 2000$  bar
  - HLF-series pumps conform to UL Class I, Div 2, Groups A, B, C, & D, T4 environments
- Air and electric valve continuous flow systems
  - Max Temp: 160 °C (air); 200 °C (electric)
  - Flow rates:  $1 \mu\text{L min}^{-1}$  to 133 mL min<sup>-1</sup>
  - Max Pressure:  $> 1300$  bar



# SEPARATORS



## Zaiput Flow Technologies



### Liquid-Liquid/Gas-Liquid Separators

- Provide continuous separation of immiscible phases by exploiting differences in wettability of a porous membrane
- Max Temperature: 130 °C; Max Pressure: 20 bar
- Wetted parts: ETFE, PFA, FEP, PTFE
- 0.5 mL internal volume; 0 – 10 ml min<sup>-1</sup> total flow rate

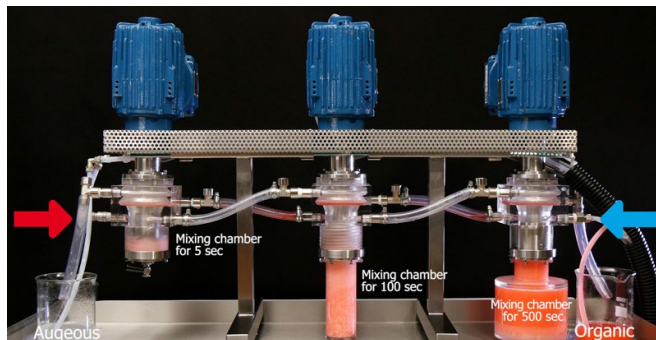
### Multi-Stage Extraction Platform

- Bench-scale tool for countercurrent liquid-liquid extraction
- Max Temperature: 80 °C
- Wetted parts: ETFE, PFA, FEP, PTFE, FFMK, PVDF
- ~3 mL per stage internal volume, 0 – 10 ml min<sup>-1</sup> total flow rate



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## CINC Centrifugal Extractors



- Continuous mixing / extraction / separation in one step
- Device can be retrofitted to increase the mixing time in a larger mixing volume
- Lab scale results easily scaled up to production
- Hastelloy C22 construction available; ATEX compliant
- Temperature –30 to +130 °C; Pressure: 20 bar
- 0 – 1000 ml min<sup>-1</sup> total flow rate

© CINC Deutschland GmbH & Co. KG



# HYDROGEN GAS GENERATORS



© ThalesNano, Inc. All rights reserved.

## ThalesNano H-Cube™ Series

- Hydrogenation without cylinders
- Electrolytic cells generate H<sub>2</sub> up to 60 NmL/min and 100 bar
- Temperature range: 10 to 150 °C
- Flow rates: 0.3 to 3 mL/min



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## ThalesNano H-Genie™ High Pressure H<sub>2</sub> Generator

- Generates ≥ 99.99% purity H<sub>2</sub>
- Gas flow range: 0.1 to 1 NL
- Pressure range: 1 to 100 bar
- Temperature range: 10 to 150 °C
- Suitable for batch, continuous, and balloon fill operation
- Monitors and records H<sub>2</sub> consumption
- Uses < 1 μS cm<sup>-1</sup> water
- Requires ion filter for H<sub>2</sub>O reservoir (consumable)



# SUMMARY AND ADDITIONAL RESOURCES



- **Advantages of flow chemistry**
  - Safety
  - Expansion of reaction space to include “forbidden” reactions
  - Lower CAPEX and OPEX
  - Reduced footprint
  - Ease of scale-up
- **Advances in continuous reactor technology have expanded the equipment capability envelope to accommodate an increasing number of process demands**

## Additional Resources

- *Flow Chemistry*, Volumes 1 & 2 by F. Darvas, G. Dormán, and V. Hessel (Eds.)
- *The Hitchhikers Guide to Flow Chemistry*, *Chem. Rev.*, **2017**, *117*, 11796–11893.
- Flow Chemistry and Continuous Processing Conference (virtual) 10 – 12 May, 2021
- Flow Chemistry Congress 2021, Boston, MA (Dates to be confirmed)
- **Questions?**
  - joseph.rheinhardt@cssquaredllc.com





# ACKNOWLEDGEMENTS



- **This work was funded by the following Prime Contract/Subcontract:**
  - Prime Contract No. W911NF-15-D-0014 Task/Delivery Order No. W911NF18F0061 from the US Army Contacting Command
  - Leidos Subcontract No. P010220377
- **The presenter would like to acknowledge the support and guidance of the following individuals:**
  - Mr. Anthony Di Stasio, Deputy Director, DPA Title III Program, Industrial Policy, OUSD (A&S)
  - Dr. Kimberly Spangler, Program Manager, Joint DoD/DOE Munitions Program OUSD (R&E)
  - Dr. Paritosh Dave, Senior Program Manager, Leidos